

Thermal Conductivity and Diffusivity of Solid/Liquid Phase Change N-Alkanes Determined by the Transient Hot Wire Technique

Catalina Velez, Mohamed Khayet^C and Jose M. Ortiz de Zarate^S
Department of Applied Physics I, Complutense University, Madrid, Spain
khayetm@fis.ucm.es

The transient multi-current hot wire technique is used to determine simultaneously the thermal conductivity (λ) and thermal diffusivity (a_T) of the solid/liquid phase change linear n-alkanes at atmospheric pressure in the range 258-348 K. These compounds are used in Phase Change Materials (PCM) for energy storage and recovery. The same set-up was used to measure λ and a_T of the liquid and the solid states at different electrical currents. Both odd- and even-numbered *n*-alkanes were considered (n-pentadecane, C₁₅H₃₂, n-hexadecane, C₁₆H₃₄, n-heptadecane, C₁₇H₃₆, n-octadecane, C₁₈H₃₈, n-nonadecane, C₁₉H₄₀ and n-eicosane, C₂₀H₄₂). The obtained results were compared with available literature data. In fact, no literature data were found for a_T of the n-alkanes considered in this study. The values obtained for the liquid phase cannot be in any reliable way extended to the solid phase. The discontinuity of λ near T_m was found to be higher for the *n*-alkane having the larger number of carbon atoms. The diffusivity a_T of C₁₆, C₁₈ and C₂₀ is larger in the solid phase than in the liquid phase. Low accuracy for both λ and a_T were obtained near (and below) the solid/liquid phase transition of the three *n*-alkanes, due to the latent heat of melting/crystallization that may affect the temperature of the wire.